## Scientists reverse notorious role of hydrogen in steel



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Hydrogen-induced ductilization in a novel lightweight TWIP steel. Credit: Hung-Wei (Homer) Yen

Hydrogen has been known to be detrimental to steels since 1875. Just a few ppm of nascent hydrogen can cause large and harmful changes in the properties of steels, known as hydrogen embrittlement (HE) or hydrogen-induced delayed fracture. In fact, many notable failures in history have

been attributed to HE. Hydrogen also limits the applications of newlydeveloped advanced high-strength steels. One classic example is twinning-induced plasticity (TWIP) steel. This steel, whose plasticity is enabled by deformation twins and dislocation glide, exhibits high strength and very large ductility. However, TWIP steel is prone to HE, and this flaw limits its potential applications, particularly in automobiles.

## Unexpected discovery from hydrogen

We have developed a lightweight TWIP steel exhibiting high plasticity. The deformation starts with dislocation glide and changes into the TWIP effect. This plastic behavior is enabled by the unique staking fault energy (SFE) and compositional complexity of the steel, i.e., the synergetic addition of Si and Al into Fe-Mn-C alloy. This novel steel is 10% lighter, strong and very ductile. However, being a TWIP steel, it can be expected to be extremely susceptible to hydrogen. Surprisingly, its mechanical properties improve after uptaking some hydrogen.

Zen-Hao Lai, one of the developers of this steel on my team, found that this steel exhibited extra ductility when it was charged with hydrogen. His first thought was that the slow strain-rate tests might have been improperly performed. After successive iterations, we confirmed the new mechanical response to hydrogen of this steel, which we dubbed hydrogen-induced ductilization (HID). We further validated the following characteristics in this new mechanism, as follows:

- 1. The trapped hydrogen is diffusional but not harmful.
- 2. HID is irrelevant to the diffusion of hydrogen.
- 3. The steel shows no signs of brittle fracture.

All these points are entirely contrary to the principle of hydrogen embrittlement. The diffusion of hydrogen in austenitic steel is so slow that the charged hydrogen should be trapped only in the subsurface of this steel. Based on theoretical calculations, the surface concentration of hydrogen can be as high as 500 ppmw. A positive impact, i.e., ductilization, from a high concentration of hydrogen was completely unexpected. We soon identified a critical question: How can hydrogen in the subsurface enhance ductility?

## A stacking fault energy of serendipity

The answer starts from the fact that hydrogen is able to reduce stacking fault energy (SFE) or increase twinnability of austenitic alloy, which has been proposed by several research teams recently. However, an appropriate SFE is required for HID. When SFE is low as known in typical TWIP steel, hydrogen eases intergranular fracture under intersections between deformation twins and grain boundaries. When SFE is too high, no ductilization effect is observed and, in some austenitic steels, hydrogen-enhanced local plasticity causes embrittlement.

In this work, published in *Scripta Materialia*, a special SFE had been achieved. Hydrogen in subsurface reduced local SFE and enabled hydrogen-enhanced densified twinning. Importantly, such twin-strengthened surface retards development of plastic localization, which invokes the typical fracture of this steel. Thus, the deformation can continue to a larger strain without much change in global work hardening behavior. The SFE of this steel is 49.5 mJm<sup>-2</sup>, a value of serendipity making HID occur. Therefore, HID is an inevitable outcome when compositional complexity was designed for this novel lightweight steel.

## Hydrogen makes steel better under twinnability engineering

In summary, elaborate twinnability engineering with an adequate SFE

not only prevents HE but brings about extra ductility, i.e. HID. To prevent HE for TWIP steel, a common method is to increase SFE by adding aluminum (> 2 wt. %). However, a positive role of hydrogen is for the first time claimed in TWIP steels. By twinnability engineering, we recently invoked HID in high-entropy alloy manufactured by selective laser melting in an unpolished work. Additionally, some similar ideas were also found in other austenitic alloys such as Fe-Al-Mn alloy, high-entropy alloy and stainless steel.

Safer, stronger, lighter and cheaper alloys for hydrogen storage and transportation are critical to net-zero society. Twinnability engineering blazes new trails toward advanced high-strength steels for storage of liquids or gaseous hydrogen. We now attempt to derive a model to predict the occurrence of ductilization. Moreover, we hope this new principle could accelerate materials development for hydrogen energy and economics.

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